

Investigation of the T4 and T6 heat treatment cycles of semi-solid processed aluminium alloy A356

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Abstract: The heat treatment cycles that are currently applied to semi-solid processed components are mostly those that are in use for dendritic casting alloys. These heat treatments are not necessarily the optimum treatments. It is shown that a decrease of the solution treatment time at 540°C from 6 hours to 1 hour does not alter the T4 or T6 tensile properties of the SSM processed A356 alloy. Slightly better impact properties are obtained with the shorter solution treatment. This could lead to time and energy savings. Natural aging prior to artificial aging causes the time-to-peak-hardness to be longer compared to the time when only artificial aging is used. Quenching after solution treatment in water at 70°C results in slightly lower tensile properties than after quenching in water at 25°C. The magnesium content of alloy A356 has a significant influence on the T4 and T6 properties of the alloy.

INTRODUCTION

Semi-solid metal (SSM) processing is a unique manufacturing method to produce near-net shape products for various industrial applications [1]. With this method, a globular microstructure is obtained in contrast with conventional liquid casting, where a dendritic microstructure is attained.

Two different SSM processes exist, namely thixocasting and rheocasting. With thixocasting, a specially prepared billet of solid material with a globular microstructure is reheated into the semi-solid range and formed. Rheocasting involves preparation of a SSM slurry directly from the liquid, followed by a forming process such as high pressure die casting (HPDC). The increased costs associated with thixocasting have resulted in rheocasting becoming the preferred semi-solid process [2].

A356 is a conventional aluminium casting alloy which is used in many applications due to its high strength-to-weight ratio. This alloy is also probably the most popular alloy used for semi-solid metal forming. This is due to its high fluidity and good “castability” [3]. These castings are generally heat treated to obtain the desired combination of strength and ductility. However, the heat treatment cycles that are currently applied to semi-solid processed components are mainly those that are in use for dendritic casting alloys. These heat treatments are not necessarily the optimum treatments, as the difference in solidification history and microstructure of

rheocast components should be considered [4]. It is known that alloy A356 responds to natural aging after the solution treatment and quench. The precipitation hardening that result from natural aging alone produces the useful T4 temper. The T6 heat treatment produces maximum strength in aluminium alloys. Unfortunately, the cost and production time associated with this treatment is substantial. Shortening the T6 heat treatment cycle would therefore have major implications on manufacturing cost and productivity [5].

In previous work by the authors [6], it was shown by means of aging curves that the optimum solution treatment time at 540°C is 1 hour. This represented the shortest possible solution treatment time to obtain complete dissolution of strengthening alloying elements, while still retaining a relatively fine microstructure. It was also shown that SSM HPDC A356 hardened significantly at room temperature after solution treatment. The time required to obtain maximum hardness at artificial aging temperatures of 160 to 190°C could be estimated using Arrhenius-type equations. Finally, it was concluded that the artificial aging response of the alloy could be increased by an instant transfer from quench to artificial aging.

The objective of this study was to characterise the T4 and T6 heat treatment cycles of rheocast alloy A356 by means of tensile and impact testing. As in the previous work, the influence of solution treatment time, natural aging time and artificial aging parameters were investigated. In addition, the effects of water quench temperature on the T4 and T6 tensile properties were also studied. Lastly, the influence of Mg-fluctuations

from 0.25wt% to 0.45wt% (the lower and upper limits of the specification, see Table 1) was also considered.

EXPERIMENTAL

Semi-solid metal slurries of A356 containing different magnesium contents (chemical composition given in Table 1) were prepared using the CSIR (Council for Scientific and Industrial Research) rheocasting process [8]. It is known that silicon has the strongest influence on the ratio of solid to liquid fraction of all alloying elements in A356 (much stronger than magnesium). A fluctuation of 1 wt% silicon in aluminium (as is allowed according to Table 1) results in a solid fraction change of almost 9% if the casting temperature is held constant [9]. For this reason, it was attempted to keep the silicon contents fairly constant, while only varying the wt% Mg. It can also be seen from Table 1 that strontium was added in all cases. The size and shape of the silicon particles are modified with additions of strontium. It is known that modified alloys undergo fast spheroidisation, while complete spheroidisation is not achieved in unmodified alloys, even after long solution treatment times. Therefore, shorter solution heat treatments can be employed with modified castings [10].

Plates (4 mm × 80 mm × 100 mm) were cast in steel moulds with a 50 ton high pressure die casting (HPDC) machine. Solution heat treatments were performed at 540°C for either 1 hour or 6 hours, followed by a water quench (25°C or 70°C). The samples were then naturally aged (NA) for various times before being artificially aged to determine artificial aging curves.

To study the T4 temper, samples were naturally aged for at least 120 hours after solution treatment and quench. Vickers hardness (VHN) was determined (20 kg load) from the average of at least four readings per sample. The average hardness values were found to be reproducible within ± 3 VHN for all heat treatment conditions tested. All samples used for microscopy were etched in 0.5% HF solution.

The tensile properties of samples in the T4 and T6 temper were determined using a total of 5 tensile tests for each heat treatment condition. The tensile samples (sub-standard size) were machined from the plates (see Fig. 1 for the dimensions of the samples).

The impact strengths were determined using a Zwick impact tester with a 40 kpcm hammer. Five non-standard size samples (55mm x 10mm x 3mm with a 45° V-notch of 2mm depth) were used for each heat treatment condition.

Table 1: Chemical composition limits (wt%) for alloy A356 [7] and the compositions used in this study

	Si	Mg	Fe	Ti	Other (Each)	Other (Tot)
Min	6.5	0.25	-	-	-	-
Max	7.5	0.45	0.20	0.20	0.05	0.15
Tensile and impact						
	Si	Mg	Fe	Ti	Sr	
	7.1	0.25	0.14	0.08	0.03	
	7.1	0.31	0.13	0.09	0.02	
	6.9	0.45	0.12	0.09	0.02	
Quench						
	7.1	0.36	0.10	0.07	0.02	
	7.2	0.40	0.10	0.08	0.02	

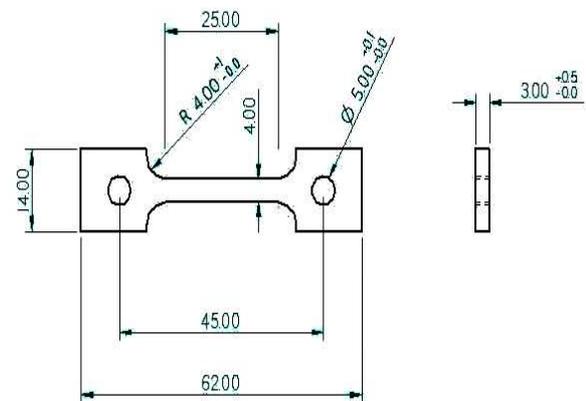


Fig. (1). The dimensions of the samples (in millimeters) used for tensile testing

RESULTS AND DISCUSSION

Fig. (2) shows a typical optical micrograph of the A356 after SSM HPDC. It is seen that the material has a globular primary grain structure and a fine eutectic.



Fig. (2). Typical optical micrograph of SSM HPDC alloy A356 (with 0.31% Mg)

SOLUTION TREATMENT AT 540°C

Figs. (3) and (4) show typical optical micrographs of alloy A356 after solution treatment at 540°C for 1 hour and 6 hours respectively. Solution treatment resulted in the eutectic structure changing to a globular type structure. It is seen that the silicon particles of the eutectic are much coarser after a solution treatment for 6 hours than after 1 hour, and that the silicon interparticle spacing is increased by the longer solution treatment.

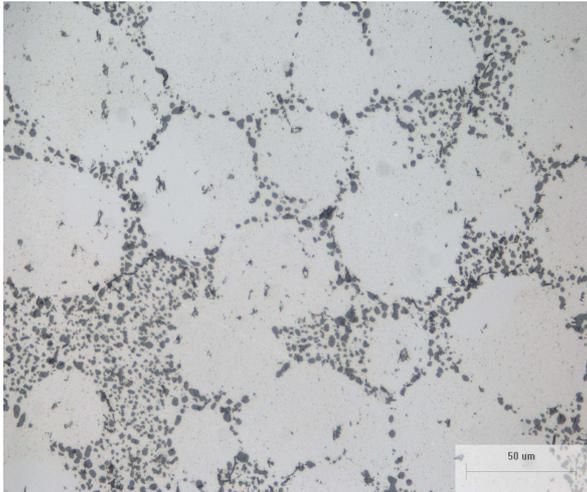


Fig. (3). Optical micrograph of alloy A356 (0.31% Mg) after solution treatment at 540°C for 1 hour

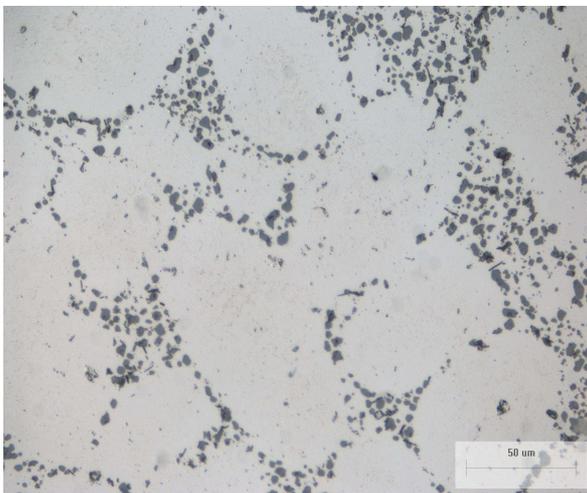


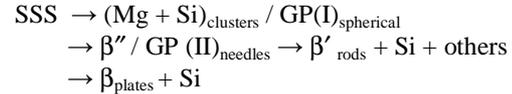
Fig. (4). Optical micrograph of alloy A356 (0.31% Mg) after solution treatment at 540°C for 6 hours

NATURAL AGING AND A356-T6

Influence of Mg-content

Fig. (5) shows natural aging curves for SSM HPDC alloy A356 after solution treatment at 540°C for 1 hour, followed by a water quench. The material is soft immediately after quenching (VHN = ~ 53), but thereafter the hardness increases rapidly. The

hardness eventually levels out after about 120 hours – the T4 temper. The magnesium content (between the lower and upper limits of 0.25 to 0.45 wt% - Table 1) has a significant effect on the natural aging behaviour of alloy A356. The reason for this can best be understood by considering the decomposition of the supersaturated solid solution (SSS) [11]:



where GP = Guinier-Preston zones

β = equilibrium Mg_2Si

β' and β'' = metastable precursors of β

The natural aging response of alloy A356 is considered to be due to (Mg+Si) clusters and GP zones. A higher %Mg should lead to a higher volume fraction of these clusters and GP-zones, resulting in improved hardness and strength.

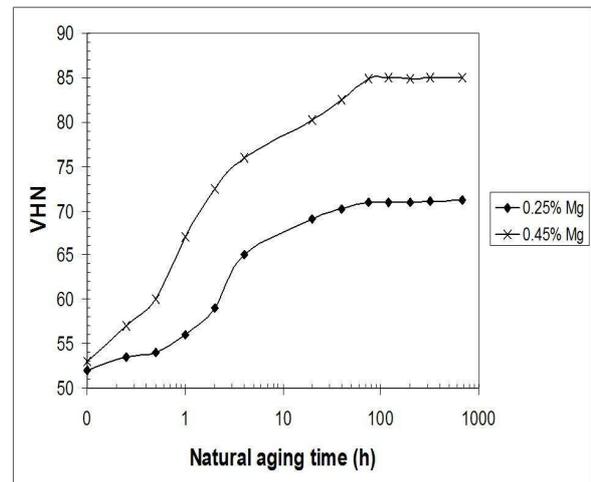


Fig. (5). Natural aging curves for SSM HPDC A356 following solution treatment at 540°C for 1 h and a water quench

Tensile and impact properties of A356-T4

The tensile and impact properties of SSM HPDC A356-T4 are shown in Table 2.

Similar tensile properties are obtained when the material was solution treated for 1 hour or 6 hours at 540°C. This implies that energy and time savings can be achieved by using the shorter solution treatment of only 1 hour. The tensile properties presented in Table 2, as well as some of the tensile properties presented later in Table 3, were used to show that linear relationships are obtained between wt% Mg and the yield strength (YS) and ultimate tensile strength (UTS) of A356-T4 (Figs. 6,7).

Slightly higher impact properties are obtained after a solution treatment of 1 h rather than 6 h for each

composition (Table 2). This is in contradiction to what was found by Zhang *et al* [12] for permanent mould cast A356-T6. They found that shorter solution treatments than the standard 6 hours resulted in lower impact strengths. This was attributed to the smaller interparticle Si-spacing after shorter solution treatments, which assisted with crack growth. However, Ogris [13] showed that the impact strength for thixoformed A356-T6 was better using only 3 minutes solution treatment compared to 12 hours. This was attributed to the fact that the Si was three dimensionally separated after 3 minutes at 540°C, whereas longer times caused the silicon to agglomerate and form large interconnected silicon crystals.

Table 2: Yield Strength (YS), Ultimate Tensile Strength (UTS), % elongation and impact strength of T4 heat treated samples. The standard deviation from five values for tensile and impact properties is also indicated in brackets.

Heat treatment	YS (MPa)	UTS (MPa)	% Elong	Impact Strength (kJ/m ²)
0.25% Mg				
540°C, 1 h	137 (4.2)	252 (6.2)	20.2 (0.94)	110.0 (3.7)
540°C, 6 h	133 (2.9)	244 (3.5)	17.2 (2.5)	106.9 (9.1)
0.31% Mg				
540°C, 1 h	146 (7.3)	263 (8.8)	17.4 (1.8)	103.0 (5.5)
540°C, 6 h	148 (4.2)	263 (6.0)	16.9 (2.3)	97.6 (6.0)
0.45% Mg				
540°C, 1 h	163 (1.5)	284 (2.6)	16.5 (1.9)	87.3 (6.4)
540°C, 6 h	167 (1.7)	286 (2.8)	17.7 (1.7)	77.1 (4.7)

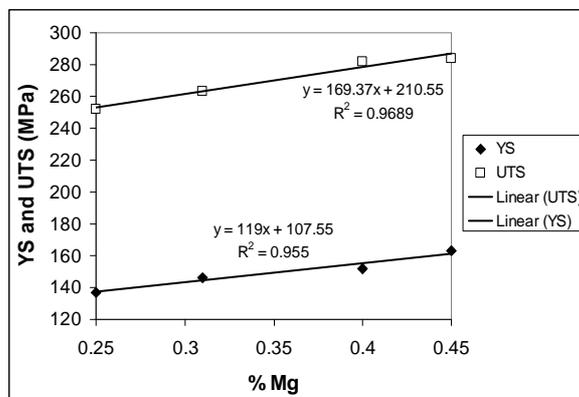


Fig. (6). Yield strength and UTS of alloy A356 as a function of the Mg-content (wt%) of the alloy for the "540-1" T4 heat treatment

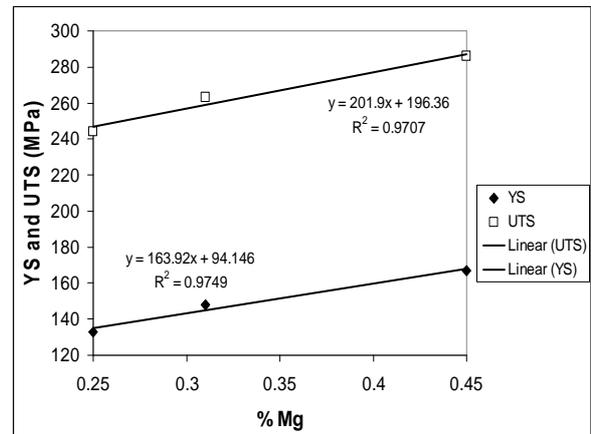


Fig. (7). Yield strength and UTS of alloy A356 as a function of the Mg-content (wt%) of the alloy for the "540-6" T4 heat treatment

The wt% Mg of alloy A356 seems to play a more important role than solution treatment time on the impact strength in the T4 temper, with high Mg-content resulting in lower impact strengths (Table 2). Linear relationships are found between impact strength and wt% Mg (see Fig. 8).

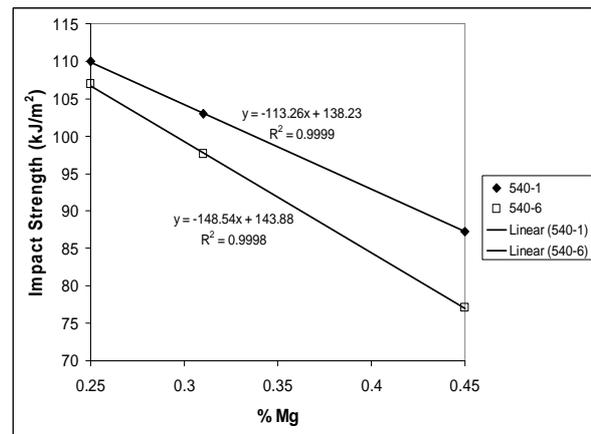


Fig. (8). Impact strength as a function of Mg-content of alloy A356-T4.

A356-T4 as a function of water quench temperature

The results presented in Table 2 were obtained by water quenching (WQ) the A356 after solution treatment in water at room temperature (25°C). However, higher quench temperatures (65-100°C) are frequently used for A356 castings to minimise distortion [10]. Less supersaturation of solutes should be obtained with a high temperature quench and the resultant T4 properties are expected to be slightly decreased [10]. To characterise this phenomenon, A356 samples were quenched in water at 25 and 70°C after the solution treatment. As expected, quenching at 70°C after solution treatment results in lower peak hardness

(Fig. 9) and tensile properties than quenching in water at 25°C (Table 3, Fig. 10).

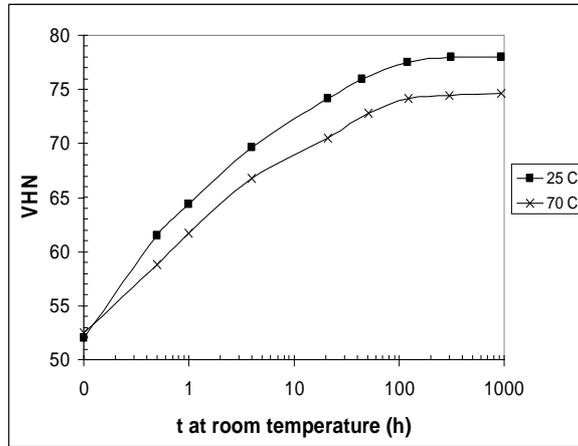


Fig. (9). Natural aging curves for 0.36wt% Mg-A356 solution treated at 540°C for 1h, followed by water quenching at 25 and 70°C

Table 3: YS, UTS, % elongation of T4 heat treated samples (0.40 wt% Mg-A356). The standard deviation from five values for tensile properties is also indicated in brackets.

Heat treatment	YS (MPa)	UTS (MPa)	% Elong
540°C-1h, 25WQ	152 (3.5)	282 (6.1)	15.8 (3.3)
540°C-1h, 70WQ	138 (2.2)	264 (6.5)	16.5 (2.5)

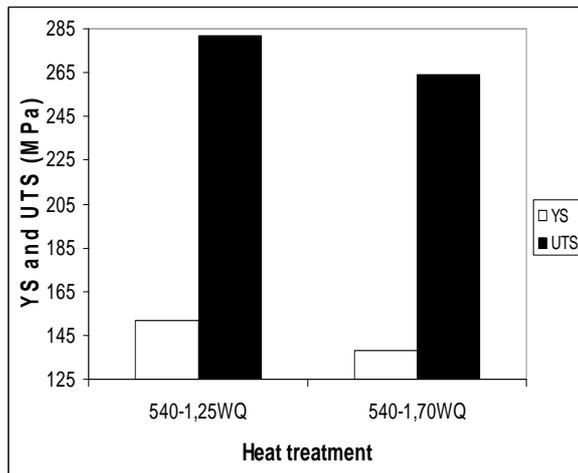


Fig. (10). A356-T4 tensile properties as a function of water quench temperature (0.40% Mg)

ARTIFICIAL AGING AND A356-T6

Influence of Mg-content and prior natural aging

The significant influence that the magnesium-content of alloy A356 also has on the artificial aging response is shown in Fig. (11) (no prior natural aging) and Fig. (12) (with 120 h prior natural aging).

Peak hardening with artificial aging (i.e. from the T6 temper) results from the precipitation of the metastable and coherent β'' [11]. In this case, a higher %Mg should, firstly, lead to a higher volume fraction of the second phase β , and most likely also of β' and β'' but, secondly, may even alter the thermodynamic and compositional characteristics of the precipitates by increasing the driving force for nucleation and thus providing a differently sized distribution.

The influence of natural aging on the subsequent artificial aging response has been studied before by the authors [6]. It was concluded that reversion of the solute clusters is responsible for the sluggish artificial aging response and initial softening in naturally aged alloy A356. It can be seen from Fig. (11) that maximum hardness is reached as a hardness plateau within 1 hour at 180°C (with no prior natural aging). If natural aging precedes artificial aging (even as short as 1 hour, as shown by the authors [6]), a hardness peak is reached after 4 hours at 180°C (Fig. 12).

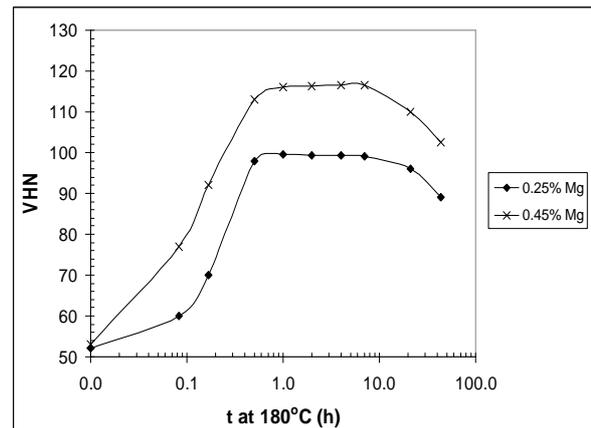


Fig. (11). Artificial aging curves at 180°C for alloy A356 (after 0 h natural aging time)

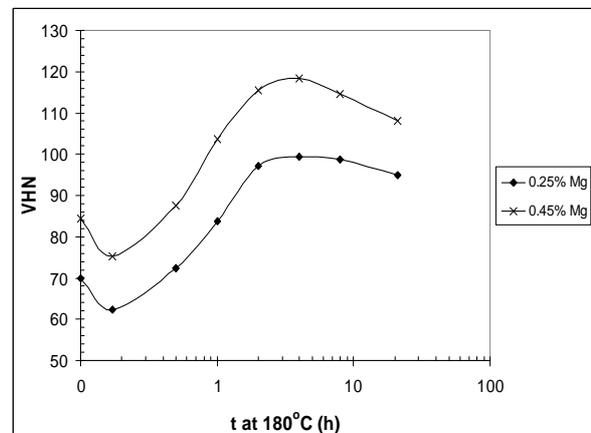


Fig. (12). Artificial aging curves at 180°C for alloy A356 (after 120 h natural aging time)

Tensile and impact properties of A356-T6

The tensile and impact properties of SSM HPDC A356-T6 are shown in Table 4.

Table 4: YS, UTS, % elongation and impact strength of T6 heat treated samples. The standard deviation from five values for tensile and impact properties is also indicated in brackets

Heat treatment	YS (MPa)	UTS (MPa)	% Elong	Impact Strength (kJ/m ²)
0.25% Mg				
540°C-6h, 0h NA, 180°C-1h	238 (1.5)	293 (6.6)	7.2 (2.5)	56.5 (6.7)
540°C-1h, 0h NA, 180°C-1h	235 (4.6)	303 (5.3)	9.7 (1.5)	57.0 (2.4)
540°C-6h, 20h NA, 180°C-4h	236 (6.8)	295 (3.7)	8.3 (1.4)	56.1 (4.1)
540°C-1h, 20h NA, 180°C-4h	243 (1.4)	303 (1.5)	7.9 (0.7)	58.3 (1.6)
0.31% Mg				
540°C-6h, 0h NA, 180°C-1h	248 (4.5)	315 (5.9)	9.3 (2.1)	50.6 (2.3)
540°C-1h, 0h NA, 180°C-1h	247 (3.2)	319 (0.6)	9.2 (0.9)	54.1 (3.0)
540°C-6h, 20h NA, 180°C-4h	255 (2.4)	317 (3.8)	9.3 (1.4)	49.1 (6.7)
540°C-1h, 20h NA, 180°C-4h	254 (6.1)	313 (7.3)	10.5 (0.9)	50.9 (4.3)
0.45% Mg				
540°C-6h, 0h NA, 180°C-1h	275 (4.2)	340 (2.2)	9.7 (1.1)	38.3 (3.8)
540°C-1h, 0h NA, 180°C-1h	277 (4.2)	342 (1.1)	8.9 (0.3)	43.0 (0.7)
540°C-6h, 20h NA, 180°C-4h	293 (3.2)	346 (1.4)	8.9 (0.8)	33.4 (2.6)
540°C-1h, 20h NA, 180°C-4h	292 (6.2)	341 (3.6)	7.8 (1.2)	34.6 (2.0)

It is seen that solution treatments of 1 hour or 6 hours at 540°C once more gave similar tensile results. The yield strength of samples that were naturally aged for 20 hours before artificial aging seems to be higher than for the samples that were not naturally aged, especially for the 0.45wt% Mg samples (Fig. 13). It has been shown by the authors [14] in subsequent work that maximum yield strength in the T6 condition (within the hardness plateau) can be achieved after 2 hours at 180°C with no prior natural aging.

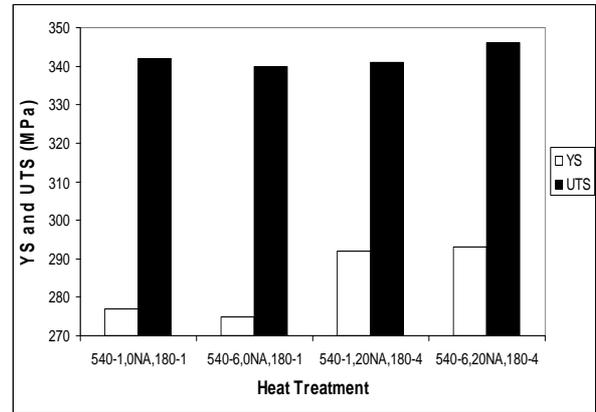


Fig. (13). Yield strength (YS) and ultimate tensile strength (UTS) of alloy A356 (0.45% Mg) as a function of heat treatment conditions.

The tensile properties shown in Table 4, together with properties obtained by the authors in other work [14-16], were combined to determine the influence of Mg-content on the tensile properties of alloy A356-T6 (see Figs. 14,15). It can be seen that linear relationships also exist between YS, UTS and % Mg for the different heat treatments.

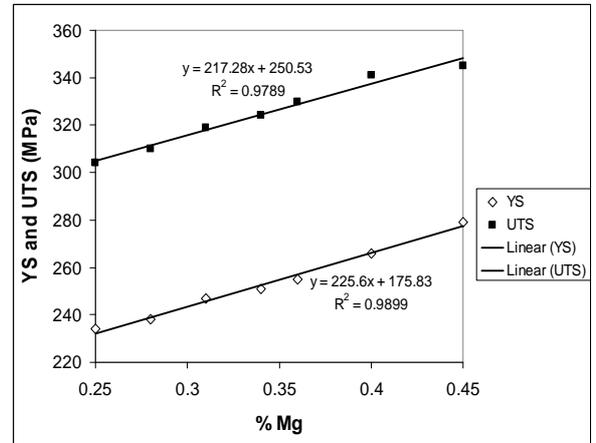


Fig. (14). YS and UTS of rheocast A356-T6 as a function of the wt% Mg of the alloy. Heat treatment: Solution treated at 540°C for 1 hour, water quenched (25°C), no natural aging, artificial aging at 180°C for 1 hour

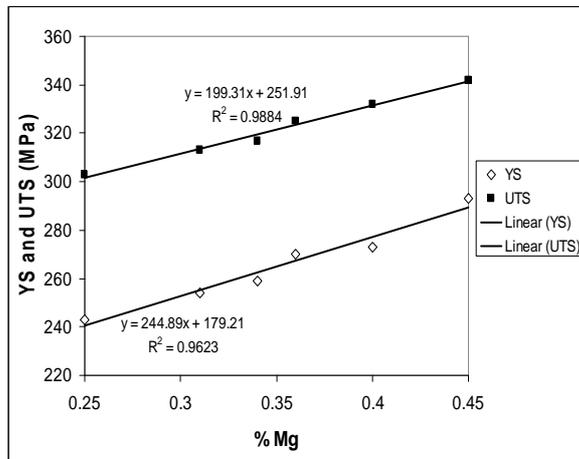


Fig. (15). YS and UTS of rheocast A356-T6 as a function of the wt% Mg of the alloy. Heat treatment: Solution treated at 540°C for 1 hour, water quenched (25°C), natural aging for 20 hours, artificial aging at 180°C for 4 hours

The impact strength of rheocast A356-T6 is much lower than A356-T4 (Tables 2 and 4). Again, the shorter solution treatment of 1 hour resulted in slightly higher impact strengths in the T6 temper (Table 4). The Mg-content of the alloy has a greater effect on the T6 impact strength than the duration of the solution treatment (see Figs. 16,17).

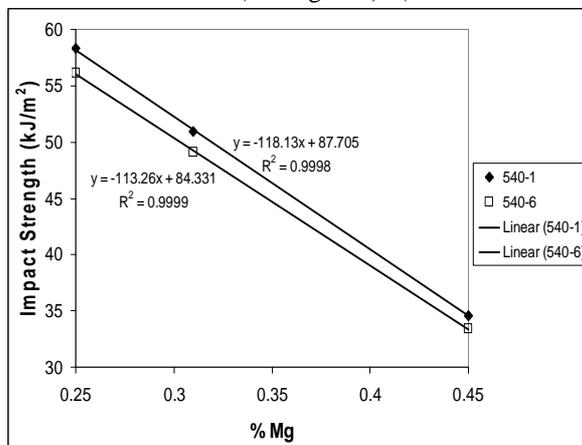


Fig. (16). Impact strength as a function of Mg-content of alloy A356-T6 (540-x,20NA,180-4).

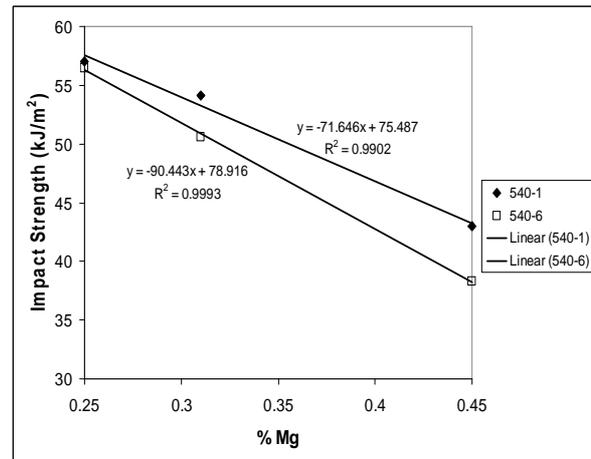


Fig. (17). Impact strength as a function of Mg-content of alloy A356-T6 (540-x,0NA,180-1).

A356-T6 as a function of water quench temperature

The results presented thus far for A356-T6 were obtained by quenching the A356 after solution treatment in water at room temperature (25°C). The influence of a higher temperature quench of 70°C on the T6 properties was also investigated (Table 5).

From Fig. (18) it is seen that the higher quench temperature leads to a more sluggish artificial aging response (presumably due to less quenched in vacancies). The peak hardness is reached after about 4 hours. This peak hardness is slightly less than when a 25°C water quench is used. This might be due to a lower degree of supersaturation obtained after the 70°C water quench. The tensile properties of samples quenched in 70°C water and artificially aged at 180°C for 1 or 4 hours are highlighted in Figs. 19-21. The tensile properties after 4 hours at 180°C are better than those after only 1 h at 180°C (Fig. 21), in agreement with the aging curves (Fig. 18).

Table 5: YS, UTS, % elongation of T6 heat treated samples (0.40wt% Mg-A356). The standard deviation from five values for tensile properties is also indicated in brackets.

Heat treatment	YS (MPa)	UTS (MPa)	% Elong
540-1,25WQ,0NA,180-1	266 (7.5)	341 (9.5)	7.8 (1.8)
540-1,25WQ,0NA,180-4	276 (8.0)	342 (4.9)	9.1 (1.2)
540-1,25WQ,120NA,180-4	280 (5.2)	337 (6.0)	6.9 (1.8)
540-1,70WQ,0NA,180-1	256 (6.0)	327 (6.7)	7.8 (0.8)
540-1,70WQ,0NA,180-4	265 (5.0)	330 (2.7)	9.8 (2.1)
540-1,70WQ,120NA,180-4	265 (3.5)	327 (4.5)	8.6 (0.7)

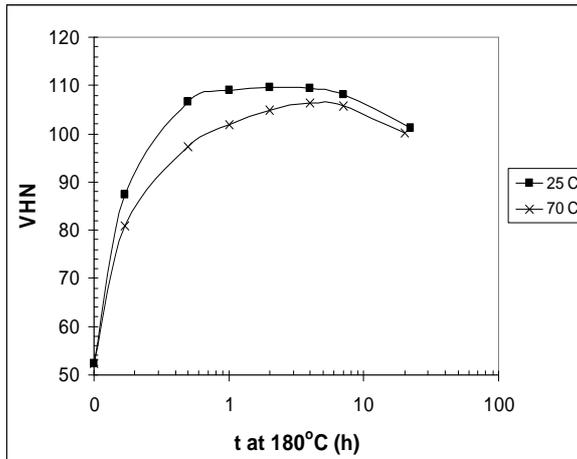


Fig. (18). Artificial aging curves at 180°C - water quench temperatures of 25 and 70°C were used and no natural aging prior to artificial aging (0.36% Mg)

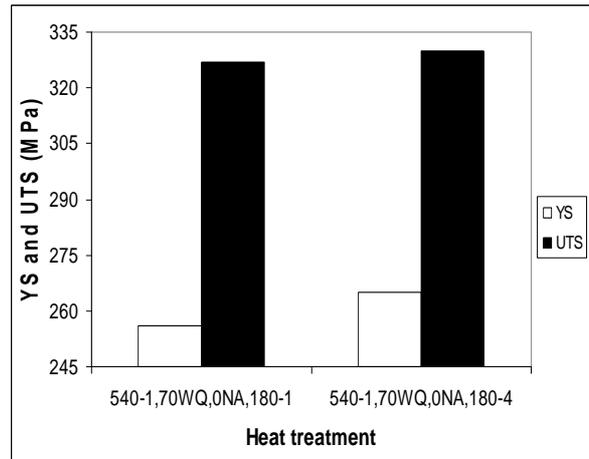


Fig. (21). Influence of artificial aging time on tensile properties of samples quenched in 70°C water. No natural aging prior to artificial aging at 180°C (0.40% Mg)

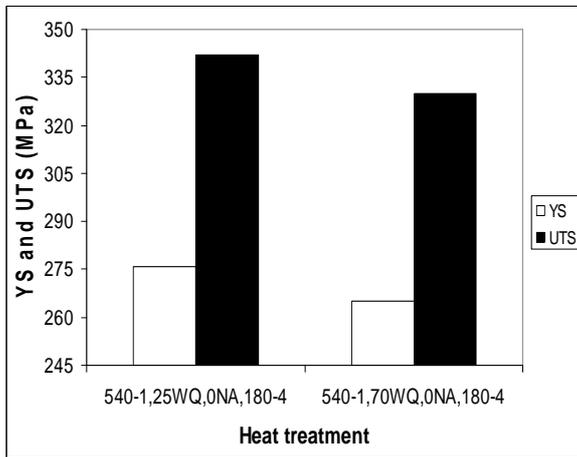


Fig. (19). Influence of water quench temperature on tensile properties. No natural aging prior to artificial aging at 180°C for 4 h (0.40% Mg)

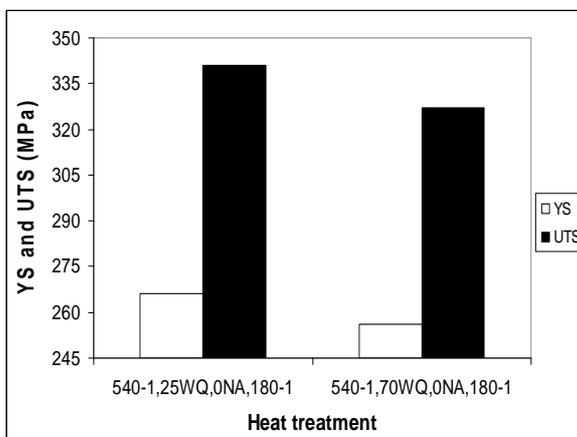


Fig. (20). Influence of water quench temperature on tensile properties. No natural aging prior to artificial aging at 180°C for 1 h (0.40% Mg)

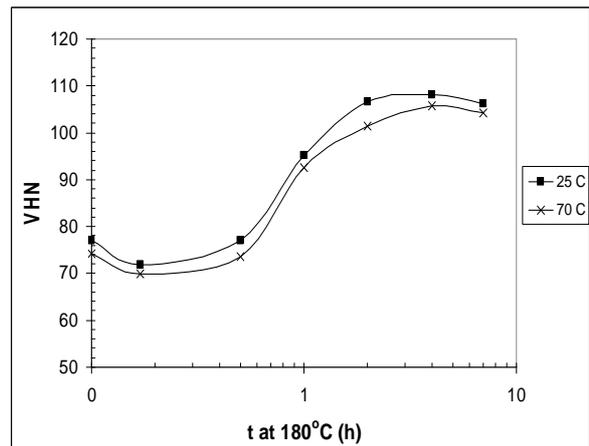


Fig. (22). Artificial aging curves at 180°C - water quench temperatures of 25 and 70°C were used and 120 h natural aging prior to artificial aging (0.36% Mg)

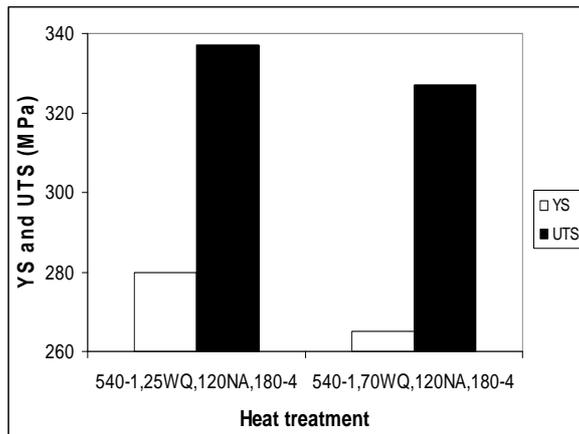


Fig. (23). Influence of water quench temperature on tensile properties. 120 h natural aging prior to artificial aging at 180°C for 4 h (0.40% Mg)

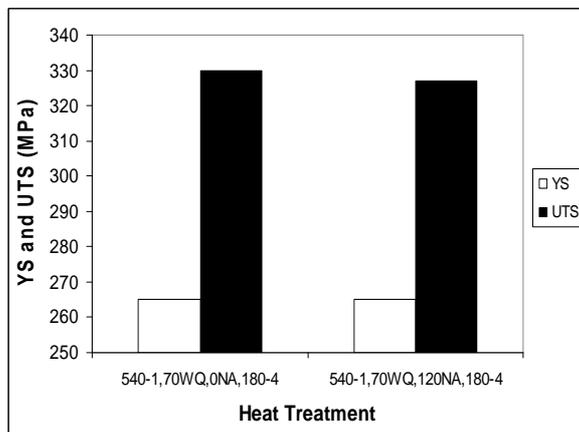


Fig. (24). Tensile properties for samples artificially aged at 180°C for 4 h after different natural aging periods (0.40 wt% Mg). Samples quenched in 70°C water.

CONCLUSIONS

The conclusions of this study are:

- The magnesium content of alloy A356 has a significant influence on the natural and artificial aging behaviour of the alloy.
- Maximum hardness is reached after only 1 hour at 180°C with no prior natural aging. This time increases to 4 hours in all naturally aged samples.
- The tensile properties of A356-T4 and A356-T6 are not influenced significantly by shortening the solution treatment from 6 hours to 1 hour.
- The impact properties of A356-T4 and A356-T6 are slightly improved by using a shorter solution treatment of only 1 hour, rather than 6 hours.

- The impact strength of SSM HPDC A356-T4 is significantly higher than that of SSM HPDC A356-T6.
- Linear relationships were found between wt% Mg of alloy A356 (within specification) and yield strength, ultimate tensile strength and impact strength for both the T4 and T6 tempers.
- Quenching in water at 70°C after solution treatment results in slightly lower tensile properties (T4 and T6) than quenching in water at 25°C.
- The higher water quench temperature after solution treatment leads to a more sluggish artificial aging response.
- The effects of any natural aging can largely be removed by an artificial aging treatment of 180°C for 4 hours, regardless of the water quench temperature after solution treatment.

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