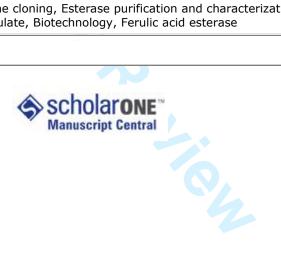
A novel recombinant ethyl ferulate esterase from Burkholderia multivorans

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1	A novel recombinant ethyl ferulate esterase from <i>Burkholderia multivorans</i>
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Aims: Isolation and identification of bacterial isolates with specific ferulic acid esteraseactivity and cloning of a gene encoding activity

Methods and Results: A micro-organism with ethyl ferulate hydrolysing activity was isolated by culture enrichment techniques. Detailed molecular identification based on species-specific primers and two conserved genes (16S rRNA and recA) led to the identification of the isolate as B. multivorans UWC10. A gene (designated estEFH5) encoding an ethyl ferulate hydrolysing enzyme was cloned and its nucleotide sequence determined. Translational analysis revealed that *estEFH5* encoded a polypeptide of 326 amino acids with an estimated molecular weight of 34.83 KDa. The EstEFH5 primary structure showed a typical serine hydrolase motif (G-H-S-L-G). The estEFH5 gene was over-expressed in E. coli in insoluble form. Following urea denaturation and *in vitro* refolding, the enzyme was purified using one-step His Select TM Nickel chromatographic column.

41 Conclusion: Purified EstEFH5 showed a preference for short chain ρ-nitrophenyl esters (C2
42 and C3) a typical feature for carboxylesterase. Furthermore, the recombinant enzyme also
43 retained the activity against ethyl ferulate.

45 Significance of the study: A biocatalytic process for the production of ferulic acid from ethyl
46 ferulate as a model substrate was demonstrated. This is the first report that describes the
47 cloning and expression of a gene encoding ferulic acid esterase activity from the genus
48 *Burkholderia*.

53 Introduction

Agro-processing industries, such as South Africa's Germiston African Products (Afprod) plant, process tons of corn per day, generating large amounts of plant cell wall waste materials. As for many other plant waste materials, such wastes have a high carbohydrate and phenolic content (Faulds et al., 1997). Such products are also potentially valuable resources for biotechnology because of the presence of high levels of hydroxycinnamic acids (ferulic, p-coumaric and p-cafferic acid) which are valuable compounds (Ishii et al., 1997). However, these phenolic acids are not readily accessible since they are typically covalently linked to polysaccharides (Saulnier et al., 1995; Kroon and Williamson, 1996). Furthermore, ferulic acid (FA) may also cross-link to form diferulic acid bridges, a process reported to be important in resistance to plant pathogens (Micard et al., 1997).

There is a growing interest from both the academic and industrial researchers to develop biotechnological processes to recover high value compounds from plant waste materials (Faulds *et al.*, 1997). The recovery of FA in particular has been of a major interest, since FA is potentially a versatile substrate for biotransformation, and can be converted with added value to vanillic acid and vanillin (Falconnier *et al.*, 1994, Kroon and Williamson, 1999, Andersen *et al.*, 2002). FA itself has a wide range of industrial applications, based both on its antioxidant properties (Graft, 1992) and its use in the food industry (Andreoni *et al.*, 1984).

Ferulic acid esterases (FAE) (EC 3.1.1.73) are a subclass of the carboxylic ester hydrolyses,
which hydrolyze ester linkages between hydroxycinnamates and sugars (Williamson *et al.*,
1998). In addition of catalyzing transesterifcation reaction (Hatzakis *et al.*, 2003), FAEs also
play an important role in hydrolyzing ethyl ferulate esters to ferulic acid (Donaghy *et al.*,
1998). These enzymes have been characterized from a number of microbial hosts including

the fungi Aspergillus niger (Christov and Prior, 1993, de Vries et al., 1997), A. awamori
(McCrae et al, 1994,) A. oryzae (Tenkanen et al, 1991), Neocallimastix MC-2 (Borneman et al, 1992), Penicillium pinophilum (Castanares et al, 1992), Butyrivibrio fibrisolvens
(Dalrymple et al, 1996) and the bacteria Streptomyces olivochromogenes (Faulds and Williamson, 1991), Pseudomonas fluorescens (Faulds et al., 1995), Bacillus and Lactobacillus sp.(Donaghy et al., 1998).

A number of studies have demonstrated ferulic acid esterase activities from different microorganisms using model synthetic substrates such as ethyl and methyl ferulate (Vafiadi *et al.*, 2006, Topakas *et al.*, 2005, Anderson *et al.*, 2002, Couteau *et al.*, 2001). Here we report the isolation, screening and identification of a *Burkholderia multivorans* isolate possessing ferulic acid esterase activity using ethyl ferulate (ethyl-3-(4-hydroxy-3-methoxyphenyl)-2propenoate) as a model substrate and the subsequent cloning and expression of a gene encoding the activity.

94 Materials and Methods

95 Unless stated otherwise reagents used in this study were supplied by Sigma Aldrich,
96 Germany. Ethyl ferulate was kindly provided by CSIR (South Africa). Minimal medium 9
97 (M9) was prepared by the method of Russell and Sambrook, (2001). M9-EF medium was
98 prepared by replacing glucose with ethyl ferulate as a carbon source.

Selective enrichment

Maize silage samples used for isolation of ethyl ferulate hydrolyzing microorganisms were
collected from a maize processing farm in Stellenbosch (South Africa). Maize silage samples
(1.0 g) were resuspended in 10 ml of sterile milli-Q water containing 2% (v/v) Tween 20.

Samples were vortexed for 5 min to dislodge microorganisms and allowed to stand for 1 h at room temperature before aliquots (100 μ l) of serially (10-10⁻⁵) diluted samples were aseptically spread onto M9-EF minimal medium plates containing various concentrations of ethyl ferulate (0.1-0.5%) as a sole carbon source. Filter sterilized cyclohexamide 0.1% (w/v) was added to the medium to inhibit fungal growth. The plates were incubated aerobically at a range of temperatures (25, 30, 37, 45 and 50 °C). Colonies were repeatedly streaked on the same minimal nutrient medium until pure cultures were obtained.

Assays

Agar assays Ethyl ferulate agar assay were prepared essentially as described by Donaghy et al. (1998). Qualitative tributyrin (Ro et al., 2004) and Olive oil-Rhodamine B (Kouker and Jaeger, 1987) agar assays were used to screen for esterase and lipase activities, respectively. **Quantitative Assays** All assays were performed in triplicate. Ethyl ferulate hydrolyzing assay was performed using method of Andersen et al. (2002). Quantitative esterase and lipase assays were performed by

measuring the release of ρ-nitrophenol as described by Petersen *et al.* (2001) and Gupta *et al.*(2002), respectively. Protein concentrations were determined by the method of Bradford,
(1976) using bovine serum albumin (BSA, Sigma Aldrich) as a standard.

143 Thin layer chromatography (TLC) analysis

After biotransformation of EF by selected isolates, supernatants were acidified to pH<2 with 145 10 N HCl, extracted twice with diethyl ether, then dried over anhydrous Na₂SO₄ and 146 concentrated by evaporation under N₂:CO₂ (80:20). Extracts were dissolved in a minimal 147 volume of methanol and chromatographed on 5 x 10 cm silica gel F₂₀ TLC plates (Merck), 148 together with commercial ferulic acid as a positive control. The mobile phase was chloroform: 149 methanol: formic acid (85:15:1 v/v/v). The spots were visualized at 335 nm and R_f values 150 were measured.

154 High performance liquid chromatography (HPLC)

Supernatants after biotransformation were analyzed with a LaChrom HPLC system equipped with UV detection and an 80-position auto-injection sampler. Separation was achieved on a Wakosil II C18 reverse phase column with a gaurd column. The detector wavelength was set at 315nm. The mobile phase was water: methanol: acetic acid (80:20:2.5 v/v/v) at a flow rate of 1 ml min⁻¹.

161 Construction and screening of *B. multivorans* UWC10 genomic library

162 Chromosomal DNA isolated according to the method of Marmur, (1963), was partially 163 digested with *Sau*3A1 and fractionated by electrophoresis on 0.7% agarose gels. The 2-8 kb 164 fraction was recovered and ligated into pUC18, which was previously digested with *Bam*HI 165 and desphosphorylated with shrimp alkaline phosphatase. The ligation products were used to 166 transform chemically competent *E. coli* DH5 α cells. Recombinants were screened on LB agar 167 plates supplemented with 0.1 mM isopropyl- β -D-thiogalactoside (IPTG), 100 µg ml⁻¹ 168 ampicillin, 1% (v/v) tributyrin and 0.1% (w/v) Gum Arabic, incubated at 37 °C.

170 Construction of expression vector

To generate the His tagged *estEFH5* gene construct, the plasmid-encoded gene was amplified
by PCR using oligonucleotides EstN and EstHis carrying *Nde*I and *Hind*III restriction sites,
respectively (Table1). The PCR product was purified and cloned into the corresponding sites
of pET22b (+), resulting in the pETEFH5_{tag} expression plasmid, which was used to transform *E. coli* BL21 (DE3) cells.

179 Protein purification

180 Cultures expressing the recombinant protein were harvested and disrupted by sonication using
181 a Sonoplus HD-070 (Bandelin, Germany) set at: 5x cycle burst, 50% max. amplitude for 2
182 min, at 30 sec interval. Cells debris was recovered by centrifugation (15 000 x g, 20 min, 4
183 °C) then washed twice with Tris-HCl buffer (50 mmol l⁻¹, pH 8.0) containing 1% Triton x100.

The insoluble inclusion bodies were resuspended in denaturing buffer (100 mmol l^{-1} , NaH₂PO₄, 100 mmol l⁻¹ Tris, 8 mol l⁻¹urea, pH 8.0) at a density of 0.2 g ml⁻¹ followed by incubation at 25 °C for 1 h with gently stirring. The clear supernatants (20 ml) obtained after centrifugation (20 000 x g; 20 min, 20 °C) were loaded onto a His PrepTM FF 16/10 column (Amersham Biosciences). The unbound proteins were washed with 3 column volumes of wash buffer (100 mmol l^{-1} NaH₂PO₄, 100 mmol l^{-1} Tris, 8 mol l^{-1} , urea, pH 7.0) at a flow rate of 0.5 ml min⁻¹. Bound proteins were refolded over 5 column volumes at a flow rate of 0.5 ml min⁻¹ with a linear gradient of a decreasing urea concentration (8-0 mol l⁻¹), generated with gradient buffer (500 mmol l⁻¹ NaCl, 20 mmol l⁻¹ Tris, 20% (v/v) glycerol, pH 7.4). The refolded bound proteins were eluted with a linear gradient of increasing imidazole concentration (0-250 mmol l⁻¹) generated with elution buffer (500 mmol l⁻¹ NaCl, 20 mmol l⁻¹ Tris, 20% (v/v) glycerol, 250 mmol 1^{-1} imidazole, pH 7.4) at a flow rate of 1 ml min⁻¹. The imidazole was removed by dialysis and the protein fraction was concentrated by membrane filtration (Centriprep YM-10, cutt-off, 10 kDa, Millipore) before analysis by SDS-PAGE according to the method of Laemmli, (1970).

DNA sequencing and sequence analysis

Sequence analyses, manipulation and annotation were performed on the Gene construction kit 2 (Textco BioSoftware, Inc.), and Bioedit (Hall, 1990) software programs. Multiple sequence alignments were performed with CLUSTALX (Thompson et al., 1997). Theoratical Mw and pI were prected using the EXPASY web site (http://www.expasy.ch) (Appel et al., 1994). Nucleotide and amino acid sequences were obtained from the following databases: GenBank, DDBJ, EMBL, SWISSPROT and PDB. Homology searches were performed using the basic nty. local alignment search tool (BLAST) (http://www.ncbi.nlm.nih.gov/BLAST/) (Altschul et al., 1997). Oligonucleotides used in this study were synthesized by Inqaba Biotech (South Africa).

229 Results

230 Isolation and screening strategies

Maize silage sample were used to isolate ethyl ferulate hydrolyzing bacteria. Samples spread on M9-EF plates were incubated aerobically at various temperatures (30-50 °C) for up to eight days. Colonies were observed only on media containing 0.1% (v/v) EF incubated at 30 °C. Re-plating of selected colonies was repeated until pure cultures were obtained, based on the light microscopic observations (data not shown). Rapid growth in the presence of the EF as a sole carbon source was taken as an indication of the presence of ethyl ferulate hydrolyzing (EFH) enzymes.

Thirteen pure bacterial isolates, selected for their ability to grow on EF as a sole carbon source, were further examined for their ability to produce EF-hydrolyzing enzyme(s). In order to confirm the presence of EFH enzyme(s) in these isolates, a qualitative agar-screening assay (Donaghy *et al.* 1998) was used. The appearance of a zone of clearance around wells containing culture supernatants was indicative of the presence of EF-hydrolyzing esterase(s).

One isolate (designated UWC 10) consistently showed high EFH activity as indicated by a
large zone of clearance. This isolate also tested positive for non-specific esterase and lipase
activities using qualitative tributyrin and olive oil plate assays, respectively. Cell fractionation
studies indicated that EFH activity was intracellularly located (data not shown). Similar
observations have been made for a number of other micro-organisms including *Bacillus*, *Lactobacillus* (Donaghy *et al.*, 1998), *E. coli* and *Bifidobacterium* strains (Couteau *et al.*,
2001)

255 Molecular Identification of isolate UWC10

In order to determine the identity of isolate UWC10, molecular identification based on the 16S rRNA gene sequence as a molecular marker was performed. Genomic DNA from strain UWC10 was used as a template for PCR amplification using the universal primer pair E9F and U1510R (Table 1). A BLAST search with the 1498 bp sequence revealed that the 16S rRNA gene of UWC10 had a very high degree of nucleotide sequence identity to the genus *Burkholderia* (Table 2). The 16S rRNA gene sequence of *B. multivorans* UWC10 was deposited in GenBank under the accession number DQ103700

Since 16S rRNA gene analysis revealed very high levels of nucleotide sequence identity (99%) to a number of *Burkholderia* species, further phylogenetic analysis was required to establish the identity of UWC10 to species level. Species–specific primers (G1, G2, SPR3, SPR4), based on a region of heterogeneity within the 16S-23S spacer sequence of members of the *Burkholderia cepacia* complex were used (Table 2).

Combinations of primer pairs (G1/G2, G1/SPR3, and G1/SPR4) were evaluated for their ability to amplify UWC10 genomic DNA. No amplification was achieved with the G1/G2 and G1/SPR4 primer pairs under the conditions used. Based on the PCR algorithm for identification of Burkholderia species (Whitby et al., 2000), the absence of a PCR product with the G1/G2 primer pair suggested that the UWC10 was neither B. cepacia nor B. concepacia, while the absence of a PCR product with G1/SPR4 suggested that UWC10 was not B. stabilis. The absence of the PCR amplicon with G1/SPR3 indicated that the organism was not B. vietnamiensis, while amplification with G1/SPR3 supported the conclusion that strain UWC10 was B. multivorans.

In order to establish conclusively that UWC10 belonged to the species *B. multivorans*, a primer pair BCR1/BCR2 (Table 1), specifically targeting the *recA* gene of *B. multivorans* (Mahenthiralingam *et al.*, 2000), was evaluated. UWC10 genomic DNA was amplified with the BCR1/BCR2 primer pair and the predicted PCR product of 1043 bp was observed. The ability of the BCR1/BCR2 primer pair to produce a strong signal under these conditions strongly suggests that UWC10 is *B. multivorans*.

Biotransformation of ethyl ferulate

Ethyl ferulate deesterification was monitored during the growth of UWC10 in M9-EF minimal medium containing 0.1% (v/v) EF. The liquid medium was inoculated with washed 10% (v/v) of a stationary phase culture pre-grown in nutrient medium. Ethyl ferulate hydrolyzing activity was confirmed by the production of ferulic acid in the culture broth (Fig. 1). The decrease in FA concentration during extended periods of growth suggests that FA was being further metabolized. Under similar conditions without inoculation of the medium, EF was very stable.

Supernatant of the cultures grown on EF-containing media were analysed by Thin Layer Chromatography (Fig. 2). A single spot (R_f =0.85) was observed at time zero (Fig. 2, lane 2). After 12 h, three spots were observed: based on the R_f values, these spots were identified as EF ($R_f = 0.85$), ferulic acid ($R_f = 0.54$) and ρ -coumaric acid ($R_f = 0.41$). After 24 h of growth the substrate was completely utilized as indicated by the absence of EF spot (Fig. 3, lanes 4 and 5). However, both ferulic acid and p-coumaric acid were still detectable. Although both ferulic acid and p-coumaric acid were still observed after 48 h of growth, an additional spot $(R_f = 0.76)$ was also observed. Based on the likely metabolic products of FA, this spot was

tentatively assigned as one of the vanillin derived compounds: i.e., vanillin, vanillic acid orvinyl guaiacol (Fig.2, Lane 6).

306 Chromatoplates were developed and sprayed with ferric chloride and ethanolic-vanillin 307 reagents (Fried and Sherma, 1996). The R_f (0.76) product failed to react with ferric chloride 308 but yielded a violet colour with ethanolic-vanillin. Such a reaction is characteristic of vinyl 309 guaiacol. Furthermore, when vanillin was included as a standard, it showed an R_f value of 310 0.68 (data not shown).

312 Library screening

Approximately 10 000 ampicillin-resistant recombinants were screened on tributyrin, olive oil-Rhodamine B and FAE plate assays. Several halo-forming clones were identified, and one highly active clone produced large halo on ethyl ferulate agar plates after three days of incubation at 37 °C was selected. The clone was designated TEND5 and its recombinant plasmid (pTEND5) isolated. This clone formed halo in the absence of the IPTG, indicating that the gene encoding esterolytic activity was being expressed in *E. coli* under the control of the intrinsic promoter.

321 Nucleotide sequence analysis of the cloned pTEND5 DNA insert

Nucleotide sequence analysis of the 2.1 kb DNA insert in the clone TEND5 revealed a single open reading frame (ORF1), predicted to encode an esterolytic enzyme. ORF1 consisted of 978 bp, commencing with an ATG start codon and extending to a TGA stop codon. Possible nucleotide sequences corresponding to typical elements of *Burkholderia* or *E. coli* promoter sequences were not found in the region upstream the predicted ORF1. However, an -AGGCsequence located 4 bp upstream the ATG start codon was presumed to be the putative

333 Analysis of deduced amino acid sequence of ORF1

A classical signature motif, the pentapeptide G-H-S-L-G motif (a.a. positions 133-137), corresponding to the G-x-S-x-G motif which harbours the catalytic serine for many esterases, lipases and other hydrolases (Jaeger et al., 1999) was observed. The serine (Ser135) within the G-H-S-L-G motif is likely to be the catalytic serine. Further amino acid sequence analysis of ORF1 revealed a PIVFVHG motif (a.a. positions 61-67) corresponding well with a putative oxyanion region (Arpingy and Jaeger, 1999) comprised of the HG dipeptide and a short hydrophobic upstream stretch. A putative oxyanion binding motif (HGDAL, a.a. positions 175-179), which corresponds well with the HXDXZ motif (Arpingy and Jaeger, 1999) (where X represents any amino acids and Z a hydrophobic residue) was also observed. In subfamilies I.1 and I.2 lipolytic enzyme, this segment is conserved as HLDEI or HILDEV, respectively and normally harbours the catalytic histidine (Arpingy and Jaeger, 1999).

346 Sequence comparison of ORF1 with other esterases

347 The deduced amino acid sequence (326 a.a.) of ORF1 was employed to search for
348 homologous proteins from protein databases. The search report revealed that the ORF1 had
349 high sequence identity (46-64%), to a number of lactone hydrolyzing esterases (Table 3).
350 ORF1 was designated EstEFH5 and the corresponding gene was designated *estEFH5*. The
351 EstEFH5 amino acid sequence was deposited in GenBank under the accession number
352 AAV97951.

In order to further analyse the primary structural features of EstEFH5, multiple sequence alignments were performed with the five closely related esterase sequences (Fig. 3). Alignment of the EstEFH5 amino acid sequence with Est1, EstF1, EstMA EstPF5 and SS3 sequences clearly demonstrates the conservation of the classical G-x-S-x-G pentapeptide signature motif (a.a. 137-141). From the alignment it could be deduced that Ser139, Asp219, and His279 form the catalytic triad, which correspond to S135, His 275, and Asp215 as the putative catalytic triad in the EstEFH5 primary structure (Fig. 3).

The HG-dipeptide (a.a. positions 70-71) typical of the oxyanion region in esterases and lipases was also conserved. Among the eight residues of the oxyanion region, three were identical and four were similar within the consensus sequence. High sequence identity between EstEFH5 and other esterases were noted at a.a. positions 223-229 (GGGLLG) and 299-301 (GHM). The functions of these motifs have yet to be elucidated.

Application of the Arpingy and Jaeger, (1999) classification scheme suggested that EstEFH5 belongs to Family V. Esterases in this family share significant homology to non-lipolytic enzymes such as epoxide hydrolase, peptidases, dehalogenases and haloperoxidases, all of which possess the α/β hydrolase fold (Arpingy and Jaeger, 1999).

378 Protein expression and purification

Analysis of the reverse primer sequence (EstR1) corresponding to the 18-nucleotide coding sequence of the *estEFH5* gene at the 3'-end revealed the presence of direct repeat sequence and showed a very high GC content (83%). From these observations we concluded that PCR amplification of the estEFH5 gene amplification would be affected by secondary structure formation and/or premature chain termination in the repeat regions. The reverse primer sequence (EstR2) with a lower GC content (66%) was designed in-frame against a non-coding sequence downstream of the TGA stop codon. As a result, the EstEFH5_{tag} protein had a predicted molecular weight of 37 KDa, due to the additional 18 amino acid residue extension at the C-terminus, including the vector encoded polyhistidine sequence (MTVRCKLAAALEHHHHHH).

Several attempts (including the removing of a His-tag, the use of the weaker tac promoter of pMS470 Δ 70 (Balzer *et al.*, 1992) and lowering the temperature during expression), were made to express EstEFH5 in a biologically active form. These attempts showed that EstEFH5 activity could be detected in the intracellular fractions although SDS-PAGE analysis showed no corresponding protein band. However, the corresponding protein band was observed in the pellet fractions, suggesting that the protein was accumulating in an insoluble form (data not shown).

In vitro refolding studies were used to recover EstEFH5 in a biological active form (Fig. 4A).
The purified EstEFH5 had a specific activity of 13.4 U mg⁻¹ against ρ-nitrophenyl acetate.
Activity staining gave a single band on the zymogram corresponding exactly with the purified protein band and confirming that no other esterases were present in the purified sample (Fig. 4B).

1		
2 3 4	403	
5 6	404	Substrate specificity
7 8 9	405	The substrate specificity of the purified EstEFH5 against different fatty acyl substrates was
10 11	406	determined using ρ -nitrophenyl esters of C2-C16 (Fig. 5). EstEFH5 selectivity was highest
12 13	407	against shorter acyl chain lengths. EstEFH5 showed maximum activity (100%) against ρ -NP-
14 15 16	408	C3 followed by ρ -NP-C2 at 74%. The hydrolytic activity of EstEFH5 decreased dramatically
17 18	409	against medium chain length esters (C6-C10). EstEFH5 was not active against p-nitrophenyl
19 20 21	410	esters with <i>n</i> -acyl chain lengths of greater than 10 (C14 and C16).
21 22 23	411	
24 25	412	Using two readily hydrolysed p-nitrophenyl ester substrates (C2 and C3), kinetic constants for
26 27 28	413	EstEH5 were calculated (Table 4) using the Michaelis-Menten non-linear regression
29 30	414	hyperbola plot in GraphPad Prism version 4.0. The K_M and V_{max} values for ρ -NP-C2 were
31 32 33	415	approximately 1.7-fold lower than that of ρ -NP-C3. Catalytic turnover (k_{cat}) for ρ -NP-C2 was
34 35	416	1.6-fold higher than that of ρ -NP-C3. Specificity constant (k_{cat}/K_M) values for the two
36 37	417	substrates suggested that ρ -NP-C3 was a marginally preferred substrate.
38 39 40	418	
41 42	419	EstEFH5 activity against ethyl ferulate was first investigated using the qualitative plate assay.
43 44 45	420	EstEFH5 showed a zone of clearance (data not shown), indicating ethyl ferulate hydrolyzing
45 46 47	421	activity. To further confirm this activity, the reaction products were analysed by HPLC.
48 49	422	HPLC traces showed the presence of free ferulic acid (data not shown) confirming that the
50 51 52	423	recombinant EstEFH5 was capable of hydrolysing ethyl ferulate.
53 54	424	
55 56 57	425	
57 58 59	426	
60	427	

 429 The primary goal of this study was to identify novel enzyme(s) capable of the conversion of
430 ethyl ferulate to ferulic acid. Plate screening assays led to the isolation of *B. multivorans*431 UWC10 which consistently demonstrated high EFH activity.

Both growing and resting cells of *B. multivorans* released ferulic acid from ethyl ferulate, which was further metabolized to vinyl guaiacol. The mechanisms of FA bioconversion are well documented and include (i) non-oxidative decarboxylation, (ii) side-chain reduction, and (iii) coenzyme-A-independent deacetylation and coenzyme-A-dependent deacetylation (Priefert et al., 2000). Based on the mechanisms of ferulic acid metabolism by B. cepacia (Andreoni et al., 1984), Ps. fluorescens (Andreoni et al., 1995) and B. coagulans (Karmakar et al., 2000), a mechanistic model of ethyl ferulate hydrolysis by B. multivorans UWC10 was postulated. Our data suggest that B. multivorans UWC10 hydrolyses the ester bond between the ethyl and ferulate groups, releasing ferulic acid. The released ferulic acid is subsequently decarboxylated to vinyl guaiacol by endogenous decarboxylase activity (Andreoni et al., 1984; 1995; Karmakar et al., 2000).

Unique features of EstEFH5 and the related lactone hydrolysing esterases (Khalameyzer *et al.*, 1999) include the presence of a 30 a.a. N-terminal extension and a very high pI value (10-11). Hydropathy plots (data not shown) indicate that the N-terminal region of EstFH was highly hydrophobic, suggesting that this sequence may anchor the enzyme in the cell membrane. It has been noted that lactone hydrolyzing enzymes possessing this sequence typically require detergents to maintain the protein in a soluble state during expression and/or purification (Khalameyzer *et al.*, 1999).

The presence of an ORF upstream the EstEFH5 encoding a putative monooxygenase was also noted (data not shown). The location of monooxygenase genes adjacent to the lactone hydrolyzing esterase genes has been previously reported in *Ps. fluorescens* (Khalameyzer *et al.*, 1999) and *Acinetobacter* sp. (Taschner *et al.*, 1988). A mechanistic model has been proposed, suggesting that esterases located adjacent to monooxygenases are involved in the hydrolysis of lactones formed by Baeyer-Villiger monoxygenases (Onakunle *et al.*, 1997).

The primary structure of EstEFH5 contains a GGGL sequence, which corresponds to the G-G-G-X (X-denoting hydrophobic residue) motif. Henke et al., (2002) have shown that the presence of the GGGX motif is linked with specificity for tertiary alcohols. All enzymes bearing this motif, including pig liver esterase, several acetyl choline esterases and an esterase from *B. subtilis* were found active towards acetates of tertiary alcohols, while enzymes bearing the more common GX motif did not catalyze the model substrates (Henke et al., 2002). It is noted however that the GGGX motif in these esterases was located adjacent to the oxyanion region, whereas in EstEFH5, this motif was located towards the C-terminus of the protein. While the significance of the positioning of the GGGX motif is unclear, the ability of these enzymes to accept bulky substrates is consistent with the activity of EstEFH towards ethyl ferulate.

The presence of predicted membrane anchor domain (Leu5-Val30) within the EstEFH5 primary structure was presumed to adversely affect protein expression in *E. coli*. This was based on the observation that the esterase activity could be detected in the intracellular fractions although no corresponding protein band could be observed on SDS-PAGE. It was therefore hypothesized that either the expressed protein was largely unfolded and inactive (presumably present as inclusion bodies) or was correctly folding but anchored to the membrane in an active form.

In order to address these questions, an expression construct lacking the 30 N-terminal amino acids corresponding to the putative membrane domain was designed. The removal of this domain resulted in complete loss of esterase activity, stressing the importance of this domain in maintaining the catalytic function or folding of this enzyme. However, the protein was still found in the membrane fraction even after the removal of the putative membrane anchor domain. Attempts to use detergents to facilitate the recovery of the enzyme from the membrane were unsuccessful.

In vitro refolding experiments were attempted to recover the EstEFH5 from the insoluble fractions. *In vitro* refolding attempts based of previously published reports were unsuccessful (Siritapetawee *et al.*, 2004). However, a series of *in vitro* refolding trials showed that glycerol was necessary to stabilise the refolding EstEFH5, and was therefore incorporated in all purification buffers.

Although the recombinant EstEFH5 esterase was shown to release FA from the model
substrate (ethyl ferulate), further development of this enzyme as a possible biocatalyst for
biotechnological production of ferulic acid would require additional engineering to enhance
the solubility of the protein.

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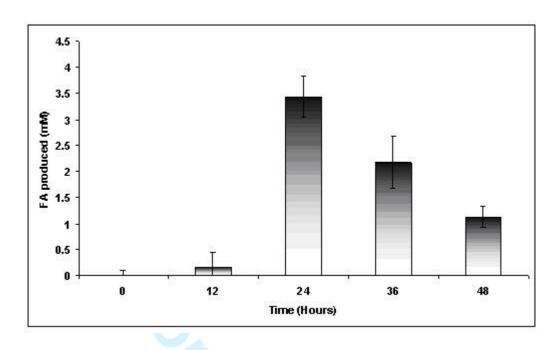
 

Figure 1: Ferulic acid produced in ethyl ferulate containing minimal medium. Data are mean values of triplicate assays. The initial EF concentration was 0.1% (w/v) (4.5 mM).

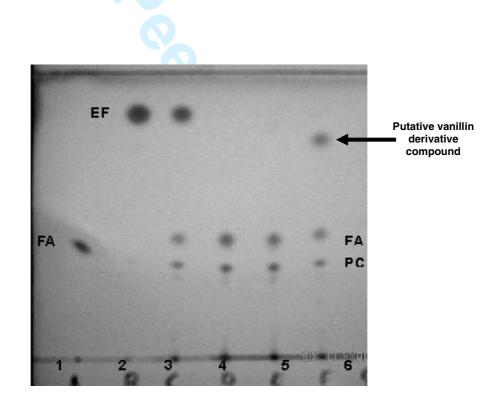


Figure 2: Thin layer chromatoplate showing ethyl ferulate (0.1% v/v) biotransformation products. Lane 1: ferulic acid $(1\mu g \mu l^{-1})$, Lane 2: 0 h, Lane 3:12 h, Lane 4: 24 h, Lane 5: 36 h, and Lane 6: 48 h. The chromatogram was developed in chloroform:methanol:formic acid (85:15:1:v/v/v) as a mobile phase and was visualized under UV light at 366 nm. FA = ferulic acid, EF=ethyl ferulate, PC=p-coumaric acid.

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Рерббо	1	HIPHIAKLI	FHEL VYAE VI	al faft qytayr	IRKHEPPEGI	EVDVGAD RLF	YVEYGEGPP:	IVFVHGL¢GQ	LRNFA
Estf1	1	HAVQWL IAAG	ILVGASV	JFWGL SAWHTRR	IEAAVPGNG	REVEVIDGEREF	YYEEGKGPP:	LVHINGLNGS	SRNL T
Est H1	1	HIQTVI	LIAVALVIA	AP VAFTFVIARR	VTKREPPEGI	GFIDIGADRVF	YTDRGQGPA	IVEVHGL¢GN	LRNFA
EstHA	1	HLISILSWLLAVLI	FLLAVLGIA	CLALATLWIAHK	AQAL VPP VGI	GFIEIDGNRIF	YVDVGEGRP:	IVFLHGL AQ	l hhf r
EstPF5	1	HEWL VARA	VFLAVSAV	JLWVVSFRINRR	IESQVPING	RFLEVGGERIF	YTDEGROPA	LTHIHELLEC	GRINL T
EstEFH5	1	HSPTLHLILMII	LLGI VAVLAJ	LALFSGYVARR	VTRAFPPEGI	RVVEVGGDRIF	YVEYG SGIP	IVEVHGLAGO	LRNFA
		90	100	110	120	130	140	150	16
							I I	I I	I
Рерббо	74	YLDL QRLAK SHRVI	LLVDRPGSGRS	er gprs sanvya	QARTIANFIA	TLGLDKP VV	GHSLG GRIS	LALALNHPQS	VSRIA
Estf1	74	YAL SRQLRE HERVI	ETLD RPGSGYS1	rrhkgtaadlpa	QARQVAAFI	401 GLDKP LV	<u>GHSLG</u> GRTS	LALALD HPEA	VSGLV
Est H1	71	YLDLERLAQ SHRVI	EVIDRPGSGRSI	.R GAASTANI YA	QARTVAQ CI (OKT GT D Ö Þ AL	GHSLG GRIR	LAVGLNHPQS	VRRLA
EstHA	79	RTLF GHF GP GYRLI	LALDRPGSGYST	TRASGTTGRLPE	QAALVRRFIN	SKLQLERP LV	GHSLG GRVT	ltlave hpea	ISGIF
EstPF5	72	HSLAPQLRERFRVI	ITID RPGSGYS1	TRARGAAADLPA	QASLVARFI	TLDLGQPLV	<u>GHSLG</u> GRVR	LSLALDHRQS	VSGLI
EstEFH5	77	YLPLARLAQHHRVI	LUDRPGAGES	er gags ganvfa	QARTIAAFII	ALRLDRP VL	GHSLG GRIA	LAVGLNHPER	VSRLA
							?		
		170	180	190	200	210	220	230	24
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PepSS3	154	LIAPLTHTETEPP	GPFR GLALRS SI	LVRRFVSLTLGL	PVAHL ON RKJ	VELVFAP EAU	PHDEGVK¢G	GLLGLOPHRE	YSRS :
Estf1	154	LVAPLTHPOPRLPI	LVFW SLAVRPAN	<i>ULRREVANTLTU</i>	PHGLLTRRSV	JUKGUFAP NAA	PEDFATRĠG	et tevrd prove	YRRS :
								GLLGINFDME	
Est H1	151	LVAPLTHNEHAPP	GAFK GLALT SPI		PLSILNSRK				YAAS :
Est H1 EstHA		LVAPLTHNE HAPP(ARRLUSWTLAU		LIAAVFAP RAF	PEDFPFKGG	GLL GLE PHVF	
	159		QRFDLLYIPSR	larrluswilau Jorwihayivai	PLSLRYARPI	liarvfap erf Thefifap Qae	PEDFPFK¢G PTDYHVE¢G	GLLGLRPHVF GWLGLRPVHF	QATSI
EstHA	159 152	LLAPLTHLE TRAR(QRFD LLYIP SRV LVFL SLAVRPAI	larrlvswtlav Jorwihaytvai Flrrwhsltlar	PL SLRYARP PHAHL GRRGI	AIAAVFAP EAF Thefifap Qae .VKSVFAP DPV	PEDFPFK6G PTDYHVE6G PEDFAER6G PHDFAVK6G	GLLGLE PHVF GWLGLE PVHF GLLGHE SDSF	QATSI YNAS:
EstHA EstPF5	159 152	LLAPLTHLETRAR(LVAPLTHPQRHLPI LIAPLSHPQSEPPI	QRFDLLYIPSR LVFL SLAVRPAI APFRPLVLPSPI	LARRLVSWTLAU JQRWIHAYTVAI ?LRRWHSLTLAR LVRRFVSWTFAI	PLSLRYARP PHAHLGRRGI PHTILTGRQJ	ATAA UFAP EAH CHEFIFAP QAF JUKS UFAP DPU AURL UFAP EDU	PEDFPFKGG PTDYHVEGG PEDFAERGG PHDFAVKGG ?	GLLGLEPHVF GWLGLEPVHF GLLGHESDSF GLLGLEPASF	QATSI YNAS: YATA 1
EstHA EstPF5	159 152	LLAPLTHLETRAR(LVAPLTHPQRHLPI LLAPLSHPQSEPPI 250	QRFDLLYIPSR LVFL SLAVRPAJ APFRPLVLPSPJ 260	LA RRLVSWTLAU JQ RWIHRYTVAI FL RRWHSLTLAR LVRRFVSWTFAI 270	PLSLRYARP PHAHLGRRGI PHTILTGRQJ 280	AIRA VFAP EAH (HEFIFAP QAE .VKSVFAP DPV AVEL VFAP EDV 290	PEDFPFKGG PTDYHVEGG PEDFAERGG PHDFAVKGG ? 300	GLLGLEPHVF GWLGLEPVHF GLLGHESDSF GLLGLEPRSF 310	QATSH YNAS: YATAJ 3:
EstHA EstPF5 EstEFH5	159 152 157	LLAPLTHLETRAR LVAPLTHPQRHLPJ LIAPLSHPQSEPPJ 250	QRFDLLYIPSRU LVFL SLAVRPAJ APFRPLVLPSPI 260 .	LA RPLUSWTLAU JQ RWIHAYTVAI FL RRWHSLTLAA LURRFUSWTFAI 270	PLSLRYARP PHAHLGRRGI PHTILTGRQJ 280	ATAAVFAP EAH THEFTFAP QAE JVKSVFAP DPV AVRLVFAP EDV 290	PEDFPFKGG PTDYHVEGG PEDFAERGG PHDFAVKGG ? 300 	GLLGLEPHVF GWLGLEPVHF GLLGHESDSF GLLGLEPASF 310	QAT SJ YNAS : YATA 1 3 :
EstHA EstPF5 EstEFH5 PepSS3	159 152 157 234	LLAPLTHLETRAR(LVAPLTHPQRHLPI LIAPLSHPQSEPPI 250 	QRFDLLYIPSR LVFL SLAVRPAJ APFRPLVLPSPJ 260 . SRYA SL SVPVD	LA RELVSWILAU JQEWIHAYIVAI FL REWHSLILAA .URREVSWIFAI 270 JL YGRGDEILWW	PL SLRYARP PHAHLGRRGI 280 .	ALAAVFAP EAH CHEFIFAP QAE .VKSVFAP DPV AVELVFAP EDV 290 	PEDFPFKGG PTDYHVEGG PEDFAERGG ? 300 	GLLGLEPHVF GWLGLEPVHF GLLGHESDSF GLLGLEPASF 310 OPAQTTDWLL	QAT SJ YNAS : YATA 7 3 :
EstHA EstPF5 EstEFH5 PepSS3 Estf1	159 152 157 234 234	LLAPLTHLETRAR(LVAPLTHPQRHLPI LIAPLSHPQSEPPI 250 	QRFDLLYIPSR LVFL SLAVRPAJ APFRPLVLPSPJ 260 . SRYA SLSVPVD KRYP QLALPIGJ	LA RRLVSWTLAU JQEWIHAYTVAI ?L REWHSLTLAA LVRRFVSWTFAI 270 III. JL YGRGDEILNW LI YGRGDEILNW	PL SLRYARP PHAHL GR GI 280 	ALRAVFAP EAH THEFIFAP QAE .VKSVFAP DPV AVELVFAP EDV 290 	PEDFPFKGG PTDYHVEGG PEDFAERGG PHDFAVKGG ? 300 G-GHHLPVI GRGHHLPII	GLLGLEPHVF GWLGLEPVHF GLLGHESDSF GLLGLEPASF 310 QPAQTTDWLL RTARVVERVL	QAT SI YNAS : YATA 7 3 : DVAGI HVAK
EstHA EstPF5 EstEFH5 PepSS3 Estf1 Est H1	159 152 157 234 234 231	LLAPLTHLETRAR(LVAPLTHPQRHLP) LIAPLSHPQSEPP 250 	QRFDLLYIPSR LVFL SLAVRPAJ APFRPLVLPSP 260 . SRYA SL SVPVD KRYP QLALPIGJ RRYA SHTVPVD	LA RRLVSWTLAU JQ EWIHAYTVAI FL RRWHSLTLAR LVRRFVSWTFAI 270 I	PL SLRYARP PHAHL GR RGI 280 	AIRAVFAP EAH THEFIFAP QAE .VKSVFAP DPV AVELVFAP EDV 290 	PEDFPFKGG PTDYNVEGG PEDFAERGG PHDFAVKGG ? 300 	GLLGLEPHVF GWLGLEPVHF GLLGHESDSF GLLGLEPASF 310 QPAQTTDWLL ATARVVEAVL QPALTTDWIL	QAT SJ YNAS : YATA 7 3 : DVAGI HVAK : GVA
EstHA EstPF5 EstEFH5 PepSS3 Estf1 Est H1 EstHA	159 152 157 234 234 231 239	LLAPLTHLETRAR(LVAPLTHPQRHLP) LIAPLSHPQSEPP 250 	QRFDLLYIPSR LVFL SLAVRPAJ APFRPLVLPSP 260 	LA RRLVSWTLAU YQ RWIHRYTVAI FL RRWHSLTLAR .VRRFVSWTFAI 270 	PL SLRYA RP1 PHAHL GR RGI PHTILT GR QJ 280 	ATAA VFAP EAH (HEFTFAP QAE , VKS VFAP DPV AVEL VFAP EDV 290 	PEDFPFKGG PEDFAERGG PEDFAERGG PHDFAVKGG ? 300 G-GHHLPVT GGGHHLPVT GUGHHPQFV	GLLGLEPHVF GWLGLEPVHF GLLGHESDSF GLLGLEPASF 310 QPAQTTDWLL QPAQTTDWLL QPALTTDWIL EPKRVIGFIE	QAT SJ YNAS : YATA 1 3 : DVAGI HVAK - GVA RIAVI
EstHA EstPF5 EstEFH5 PepSS3 Estf1 Est H1	159 152 157 234 234 231 239 232	LLAPLTHLETRAR(LVAPLTHPQRHLP) LIAPLSHPQSEPP 250 	QRFDLLYIPSR LVFL SLAVRPAJ APFRPLVLPSPJ 260 . SRYA SL SVPVD KRYP QLALPIGJ RRYA SHTVPVD QRYGEIVHPAGJ KHYP SLKLPVGJ	LA RELVSWTLAU JQEWIHAYTVAI FL REWHSLTLAR LVREFVSWTFAI 270 I	PL SLRYARP PHAHL GR EGI PHTILT GR QJ 280 QRHGEAL KKI RRHGQALADI KRQGEAL KQI RIHGEPHRGI RIHGE SHVGI	ALRAVFAP EAH THEFIFAP QAE .VKSVFAP DPV AVEL VFAP EDV 290 	PEDFPFKGG PTDYHVEGG PEDFAERGG ? 300 G-GHHLPVT GEGHHLPVT GUGHHPQFV GEGHHLPIS	GLLGLEPHVF GWLGLEPVHF GLLGHESDSF GLLGLEPASF 310 QPAQTTDWLL ATARVVEAVL QPALTTDWIL EPKRVIGFIE AVERVVAHVQ	QAT SJ YNAS : YATA 1 3 : DVAGI HVAGI HVAS -

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		····I····I····I····I···	
PepSS3	313	PARAEPEHTRESVES	327
Estf1	312	RVRPVETATVLHPPFALANK-	332
Est H1	307	AAVPIQAEAAASA	320
EstHA	318	-AFPSPQPHDNFNEPTG	334
EstPF5	310	QAAPQRSATILHPPFAASRAS	331
EstEFH5	313	AAPVTAAAARAAH	326

Figure 3: Multiple sequence alignment of *B. multivorans* esterase EstEFH5 (AAV97951) and
other related proteins represented by SS3 from *B. pseudomallei* (YP11976), Estf1 *Ps. fluorescens* DMS01016 (AAC36352), Est1 from *R. mannitoliltyica* M1 (AAQ83881), EstMA
from *Mesorhizobium loti* (BAB47791) and EstPF5 from *Pseudomonas* sp. (AAY92206).
Catalytic triad residues are indicated by solid triangle. Motifs referred to in the text are boxed.

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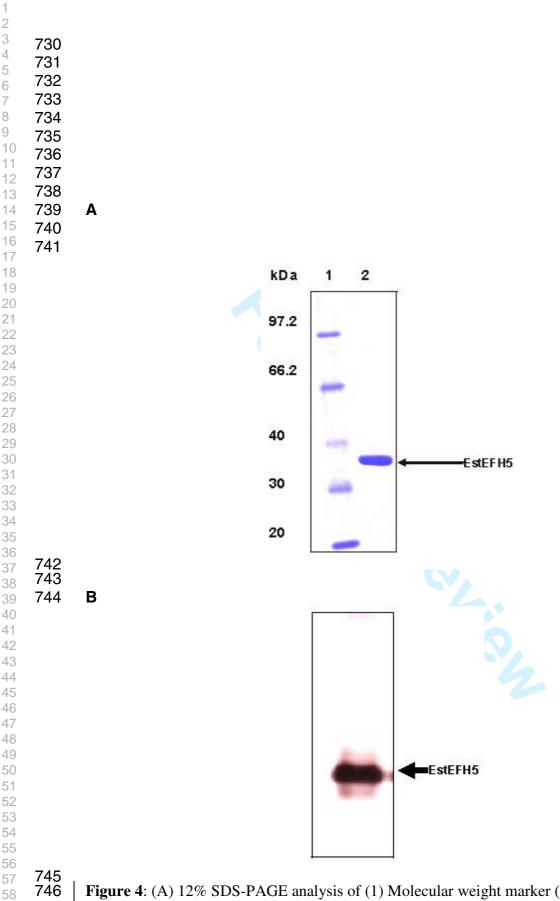


Figure 4: (A) 12% SDS-PAGE analysis of (1) Molecular weight marker (2) purified EstEFH5
and (B) zymogram assay showing activity stained EstEFH5.

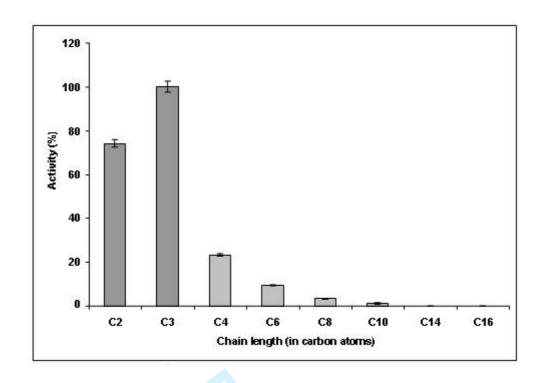


Figure 5: Hydrolysis of ρ-nitrophenyl fatty acid esters by EstEFH5 esterase (activity against ρ-NP-C3 was taken as 100%).

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Prime	r Target	Sequence (5′→3′)	References
name	- Turger		
E9F	7-26*	GAGTTTGATCCTGGCTCAG	Farrelly et al., 1995
U150U	JR 1490-1512*	GGTTACCTTGTTACGACTT	Farrelly et al., 1995
G1	153-133†	GCCATGGATACTCCAAAAGGA	Whitby <i>et al.</i> , 1998
G2	939-958†	TCGGAATCCTGCTGAGAGGC	Whitby <i>et al.</i> , 1998
SPR3	969-985 §	TCGAAAGAGAACCGGGCG	Whitby <i>et al.</i> , 2000
SPR4	939-985 §	TCGAAAGAGAACCGATA	Whitby et al., 2000
BCR1	(F) 2-20**	TGACCGCCGAGAAGAGCAA	Mahenthiralingam et al., 2000
BCR2	(R) 1044-1024**	CTCTTCTTCGTCCATCGCCTC	Mahenthiralingam et al., 2000
EstR1	<mark>963-981^{††}</mark>	TCAGTGCGCCGCGCGCGC	This study
EstR2	<mark>962-993^{††}</mark>	GCAGCGCACCGTCATGTG	This study
<mark>Est (N</mark>) <u>1-24^{††}</u>	ATCATA <u>CATATG</u> AGCCCGACGCTC	This study
		CATCTGATC	
<mark>Est (H</mark>	is) 962-993 ^{††}	ATTCGC <u>AAGCTT</u> GCAGCGCACCGT	This study
33 * = Po	tite of the sector is a sector	CATGTG coli 16S rRNA gene (GenBank Accession nun	1
		cepacia 16S rRNA gene (GenBank Accession hun	
-		nultivorans 16S rRNA gene (GenBank Accession	,
		tabilis 16S rRNA gene (GenBank Accession 1	
U I		multivorans recA gene (GenBank Accession r	
		multivorans EstEFH5 gene (GenBank Acces	

Table 2: BLAST (n) search data for the UWC10 16S rRNA gene sequence.

Strains	Accession	Score	Identity (%)	Evalue
	number	(Bits)		
B. multivorans	AF097531	2927	1489/1494 (99%)	0.00
B. multivorans	BMU18703	2919	1476/1478 (99%)	0.00
B. vietnamiensis	AF097534	2896	1485/1494 (99%)	0.00
B. multivorans	AB092606	2882	1466/1771 (99%)	0.00
B. cepacia	AB110089	2872	1479/1490 (99%)	0.00
B. cepacia	AY741359	2870	1478/1489 (99%)	0.00
B. cepacia	AY741345	2866	1464/1471 (99%)	0.00
Burkholderia sp.	AY769904	2866	1464/1471 (99%)	0.00
<i>Burkholderia</i> sp	AY973819	2866	1464/1471 (99%)	0.00
<i>Burkholderia</i> sp.	AF219125	2866	1464/1471 (99%)	0.00

798

Table 3: Comparison of the amino acid sequence similarities between EstEFH5 and other

800 related proteins

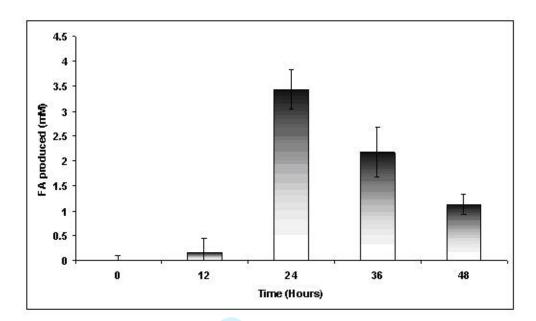
Esterase producing organisms	Accession	Identity
	number	(%)
Lactone hydrolyzing esterases group		
Burkholderia pseudomallei (SS3)	Q63IU6	64.0
Ralstonia mannitolilytica (Est1)	Q6EIRO	61.3
Pseudomonas sp. (EstPF5)	Q9AE76	44.3
Mesorhizobium loti (EstMA)	Q98HN0	36.6
Pseudomonas fluorescens (EstF1)	O87637	46.0

801		
802		
803		
804	Table 4: Kinetic parameters for hydrolysis of p-	nitrophenyl esters by EstEFH5 esterase
	ρ-NP-C2	ρ-NP-C3

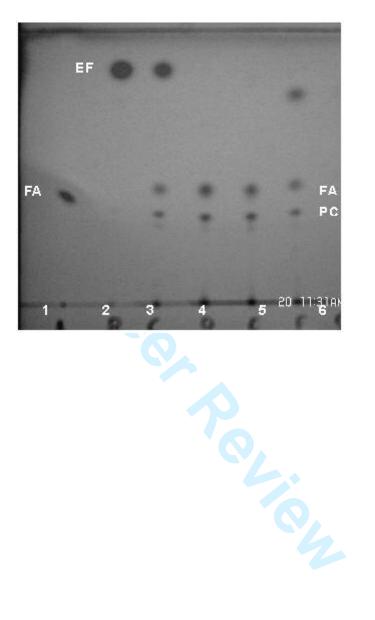
K _M (mM)	9.2 ± 1.2	5.3±.8	
V_{max} (U.mg ⁻¹)	198.9 ± 13.2	125 ± 9.0	
$k_{cat}(S^{-1})$	94.7 ± 6.3	59.5 ± 4.3	
$k_{cat}/K_M (s^{-1} mM)$	10.3	11.2	

805 Enzyme concentration in the assay= 3 μg ml⁻¹

806 K_{cat} was calculated assuming a molecular weight of 35 kDa as estimated SDS-PAGE analysis, and a single site per monomeric protein.



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36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53	
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36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55	
36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56	
36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57	
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PepSS3	1	HIPHIAKLFHFL VYAI	?VAL FAFT QYTAYR	IRKHEPPEGR	FVDVGADRLHY	VEYGEGPPI.	VEAHET COOL	RNFA 7
Estf1	1	HAVQWLIAAGVLVGAS	VVFWGL SAWHTRR	IEAAVP GNGR	FVEVDGERFHY	YEEGKGPPL	VHINGLNGSS	RNLT
Est H1	1	HIQTVLIAVALVI	AAP VAFTFVIARR	VTKREPPEGR	FIDIGADRVHY	TDRGQGPAI	VEVHGLCGNL	RNFA '
EstHA	1	HLISILSWLLAVLFLLAVLGI	ACLALATLWIAHK	ROALVPPVGR	FIEIDGNRIHY	VDVGEGRPI	VFLHGL ROL	HHFR. '
EstPF5	1	HEWL VAARVFLAVS	AVLWVVSFRINRR	IESQVPINGE	FLEVGGERINY	TDEGRGPAL:	LHINGLTGCG	RNLT
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Estf1	74	YAL SROLRE HERVITLD RPG:	GYSTRHKGTAADLPA	QARQVARFIN	QLGLDKPLVLG	HSLG GRISL	ALALD HPEAV	SGLV :
Est H1	71	YLDLERLAQ SHRVIVID RPG:	GRSLR GAASTANIYA	QARTVAQCIQ	KLGLDQP VLVG	HSLG GATAL	AVGLNHPQSV	RRLA :
EstHA	79	RTLF GHF GP GYRLIALD RP G:	GYSTRASGTTGRLPE	QAALVRRFIE	KLQLERPLVVG	HSLG GAVTL	FLAVE HPEAT	SGIA :
EstPF5	72	HSLAPQLRERFRVITIDRPG:	GYSTRARGAAADLPA	QASLVARFIE	TLDLGQPLVLG	HSLG GAVAL	SLALDHRQSV	SGLI :
EstEFH5	77	YLPLARLAQHHRVILVDRPG	GRSTR GAGSQANVFA	QARTIAAFID	ALRLDRP VLVG	HSLG GATAL	AVGLNHPERV	SRLA :
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PepSS3	154	LIAPLTHTE TEPP GPFR GLAI	.RSSLVRRFVSLTLGL	PVAHL ON RKA	VELVFAP RAVP	HDFGVKGGG	LLGLOPHRFY	SASS
Estf1	154	LVAPLTHPQPRLPLVFW SLAV	RPAWL RREVANTLTV	PHGLLTRRSV	WKGVFAP DAAP	EDFATRGG	LL GHE PDNEY.	AASS :
Est H1	151	LVAPLTHNE HAPP GAFK GLAI	TSPLARRLVSWTLAU.	PLSILNSRKA	IAAVFAP EAHP	EDFP FK¢GG	LLGLEPHVEY	AASS :
EstHA	159	LLAPLTHLETRARQRFDLLY	PSRVQRWIHRYTVRI	PLSLRYARPT	HEFIFAP QAFP	TDYHVE¢GG	VLGLE PVHFQ	ATSA :
EstPF5	152	LVAPLTHPQRHLPLVFL SLAV	RPAFL RRWHSLTLAA	PHAHL GRRGL	VKSVEAP DPVP	EDFAER¢GG	LL GHE SD SFY	NASS
EstEFH5	157	LIAPL SHPQ SEPPAPERPLVI	.PSPLVRRFVSWTFAI	PHTILTGROA	VRLVFAP EDVP	hdfa vk¢gg	LLGLEPASEY	ATAT :
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PepSS3	234	dlvaapedl pdhe srya sl sv	7PVDVLYGRGDEILNW	QRHGERL KKR	ldavnlt vvdg	-GHHLPOID	PAQTTDWLLD	VAGK :
Estf1	234	EIAL UND CL P GHVKRYP QLAI	.PIGLIYGRQDKVLDF	RRHGQALADE	VP GLKLQ VVE G	RGHHLPITA	TARVVEAVLH	VAK-
Est H1	231	dlvaapedl pdherrya shtv	/PVDVLYGRGDRILNV	KRQGEAL KQR	LERVNLRVVDG	GHHLPVIQ	PALTTDWILG	VA :
EstHR	239	DVVAVEEDL GRIEQRYGEIVI	¢PAGIL FGTADRVIDI	RINGEPHRGE	IAGLDFEPVDG	VGHHPQFVE	PKRVIGFIER	IAVR
EstPF5	232	ELAVVNRSL PGHVKHYP SLKI	.PVGLI YGSEDPVL SY	RRHGESHVGR	WPGLSLEIVQG	GHHLPI SA	VERVVRHVQH	VAS - 3
EstEFH5	237	DLL CAPDDL PAHE SRYA ELAV	7P VDVL YGREDRILDW	RANGDALAKR	SARVELT VVEG	GHHLPVTH	PEATADWLLE	VA 3
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PepSS3	313	PARREPENTRESVES	327					
Retf1	310		W- 330					

PepSS3	313	PARAEPEHTRESVES	327
Estf1	312	RVRPVETATVLHPPFALANK-	332
Est H1	307	AAVPIQAEAAASA	320
EstHA	318	-AFPSPQPHDNFNEPTG	334
EstPF5	310	QAAPQRSATILHPPFAASRAS	331
EstEFH5	313	AAPVTAAAARAAH	326

kDa 97.2 66.2 EstEFH5

